# COMPARATIVE MOLECULAR FIELD ANALYSIS OF THE [3, 3] SIGMATROPIC REARRANGEMENT

## Soumendranath Bhakat

Nutanpalli and P.O: Suri, Dist: Birbhum, 731101 West Bengal, INDIA e-mail: soumendranath2009@gmail.com

(Received May 28, 2010)

**ABSTRACT**. A comparative molecular field analysis (CoMFA) on the substituent effect of the palladium (II) catalysed sigmatropic rearrangement [3, 3] of esters was studied to show a good correlation between electrostatic property of the substituents and the reaction rate. The results suggest that the fact that the reaction rate will increase as the electron donating ability of the substituents increase or in other word the more powerful the electron donating group, the more faster the reaction proceed.

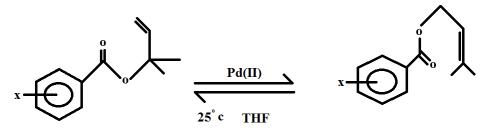
**Keywords:** CoMFA, QSAR, PLS method, [3, 3] signatropic rearrangement, ZODIAC 0.65, MOPAC 2007

## **INTRODUCTION**

Comparative molecular field analysis (CoMFA) a 3D-QSAR (Quantitative Structure Activity Relationship) concept was developed by CRAMER *et al.* [1] in 1988 and since then it became a popular and valuable tool in drug design and molecular modelling [2]. A traditional QSAR method requires predetermined parameters representing the physical and chemical properties of the molecule which is mainly derived empirically and sometimes difficult to get. On the other hand the CoMFA method only requires fundamental properties of the molecules, steric, electrostatic properties which can be obtained by theoretical calculations. So the CoMFA method offers clear advantage over the conventional 3D-QSAR. This QSAR technique mainly used in drug design can also be used to find the relationship between molecules and physiochemical property expressed by the molecule. Moreover the new CoMFA method can be used to study substituent effect on [3, 3] signatropic rearrangement.

### **RESULTS AND DISCUSSION**

Geometries of the molecule were generated by the molecular modelling suite ZODIAC 0.65 version and geometry optimization and the charge calculations were carried out using MOPAC 2007 [3] by Sparkle method. The optimized molecules are represented in the scheme 1. Then the CoMFA analysis was performed using the quantitative calculation in ZODIAC 0.65.



#### Scheme-1

The CoMFA grid spacing was 2.0 angstrom in all three dimensions (x, y, and z) and the greed is generated automatically by the programme and was large enough to contain completely with addition 4.0 angstrom-5.0 angstrom in all directions. As probes sp<sup>3</sup> C<sup>+</sup> and H<sup>+</sup> ion were used.

| Substrate | Substituent         | $k_{x}$ (min <sup>-1</sup> M <sup>-1</sup> ) | logk' |
|-----------|---------------------|--|-------|
| 1a        | p-CH <sub>3</sub> O | 130  | 0.31  |
| 1b        | p-CH <sub>3</sub>   | 92.9   | 0.17  |
| 1c        | Н                   | 63.2   | 0     |
| 1d        | p-Cl                | 53.1   | -0.07 |
| 1e        | p-Br                | 46.2   | -0.14 |
| 1f        | m-F                 | 28.5   | -0.35 |
| 1g        | m-Br                | 32.8   | -0.28 |
| 1h        | m-CF <sub>3</sub>   | 17.7   | -0.55 |
| 1i        | p-CF <sub>3</sub>   | 19.4   | -0.51 |
| 1j        | m-NO <sub>2</sub>   | 10.1   | -0.80 |
| 1k        | p-NO <sup>2</sup>   | 11.1   | -0.75 |

Table 1. - The Second order rate constant [4]

 $Logk'=logk_1/k_H$ 

A statistical analysis of the interaction energy and the target property (logk') was carried out by the partial least square (PLS) method [5] with leave-1-out, cross validation [6]. The final CoMFA model was calculated using no cross validation with an optimum number of components from the cross validation results.

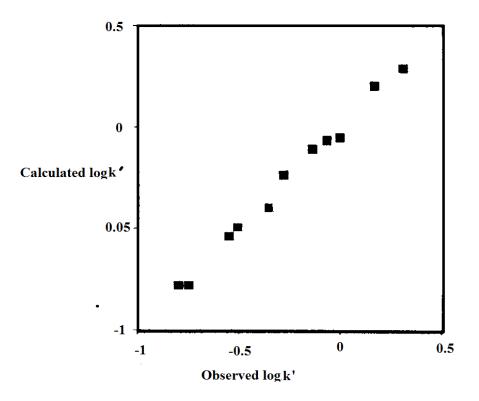
When  $C^+$  and  $H^+$  were used as probes and both steric and electrostatic models were considered (M-1, M-2, M-3, andM-5), the cross validated  $r^2$  values ( $r^2_{cross}$ :0.698, 0.713, and

0.663) were generally quite high (Tab. 2). In M-2 in which only electrostatic field is considered gave higher cross validated  $r^2$  values 0.898 and non-validated  $r^2$  values 0.991. That probe the electrostatic factor is the major factor on the reaction rate of [3, 3] signatropic rearrangements (the graph indicated this).

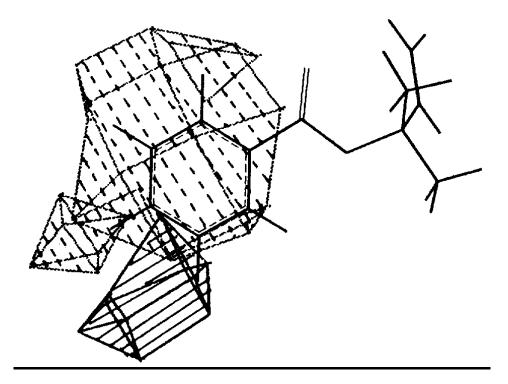
|  | M1            | M2             | M3            | M4             | M5            |
|--|---------------|----------------|---------------|----------------|---------------|
| Probe Atom                                 | $C^+$         | C <sup>+</sup> | $C^+$         | $\mathbf{C}^+$ | $C^+$         |
| Field                                      | Steric        |                | Steric        |                | Steric        |
|  | Electrostatic | Electrostatic  | Electrostatic | Electrostatic  | Electrostatic |
| Energy cut-off                             | 30/30 kcal.   | 30 kcal.       | 5/30 kcal.    | 5 kcal.        | 30/30 kcal.   |
| <b>R</b> <sup>2</sup> <sub>cross-val</sub> | 0.698         | 0.898          | 0.713         | 0.825          | 0.663         |
| No. of component                           | 2             | 3              | 2             | 3              | 3             |
| Relative contribution                      |               |                |               |                |               |
| Steric                                     | 0.323         |                | 0.297         |                | 0.317         |
| Electrostatic                              | 0.677         | 1              | 0.733         | 1              | 0.683         |
| R <sup>2</sup> <sub>no-val</sub>           | 0.966         | 0.991          | 0.947         | 0.989          | 0.972         |
| Standard error                             | 0.074         | 0.037          | 0.093         | 0.046          | 0.072         |

Table 2. CoMFA-PLS analysis of Model 1, Model 2, Model 3, Model 4, Model 5

The CoMFA method model for M-2 (Figure) shows that more negative charge around the phenyl ring will increase the target property; indicate that the electron donating group on the benzene ring will increase reaction rate.



Graph. Plot of logk' calculated using Model-2 versus observed logk'



Electrostatic model by Model 2. Dot line indicates regions where a more negative electrostatic interaction would improve the rate constant. Real lines indicate regions where a more positive electrostatic interaction would enhance the rate constant.

In conclusion it can be said that the CoMFA method is a powerful technique in analysing the kinetic data of [3, 3] sigmatropic rearrangement.

#### Acknowledgement

I am very much thankful to Dr. Roald Hoffmann of Cornell University for his comments and to Dr. Somnath Bhakat, Mr. Ashis Chakraborty and Mr. Shasti Kinkar das for their constant inspiration.

## **References:**

- [1] R. D. CRAMER III, D. E. PATTERSON, J. D BUNCE, J. Am. Chem. Soc. 110 (1988) 5959.
- [2] a) J. P. HORWITZ, I. MASSOVA, T. E WIESE, A. J. WOZNIAK, T. H. CORBETT, J. S. SEBOLT LEOPOLD, D. B. CAPPS, W. R. LEOPOLD, *J. Med. Chem.* **36** (1993) 3511.
  - b) K. H. KIM, Y. C. MARTIN, J. Med. Chem. 34 (1991) 2056.
  - c) K. H. KIM, Med. Chem. Res. 1 (1991) 59.
  - d) K. H. KIM, Y. C. MARTIN, J. Org. Chem. 56 (1991) 2723.
- [3] J. J. P. Stewart, MOPAC version 6 (QCPE no. 455). The optimizations were carried out using EF option.
- [4] K.-W. CHI, E.-C. KOO, Bull. Korean. Chem. Soc. 15 (1994) 98.
- [5] S. HELLBERG, M. SJOSTROM, B. SKAGERBERG, Wold. S. J. Med. Chem. 30 (1987) 1126.
- [6] R. D. CRAMER III, J. D. BUNCE, D. E. PATTERSON, Quant. Struct. Act. Relat. 7 (1988) 18.